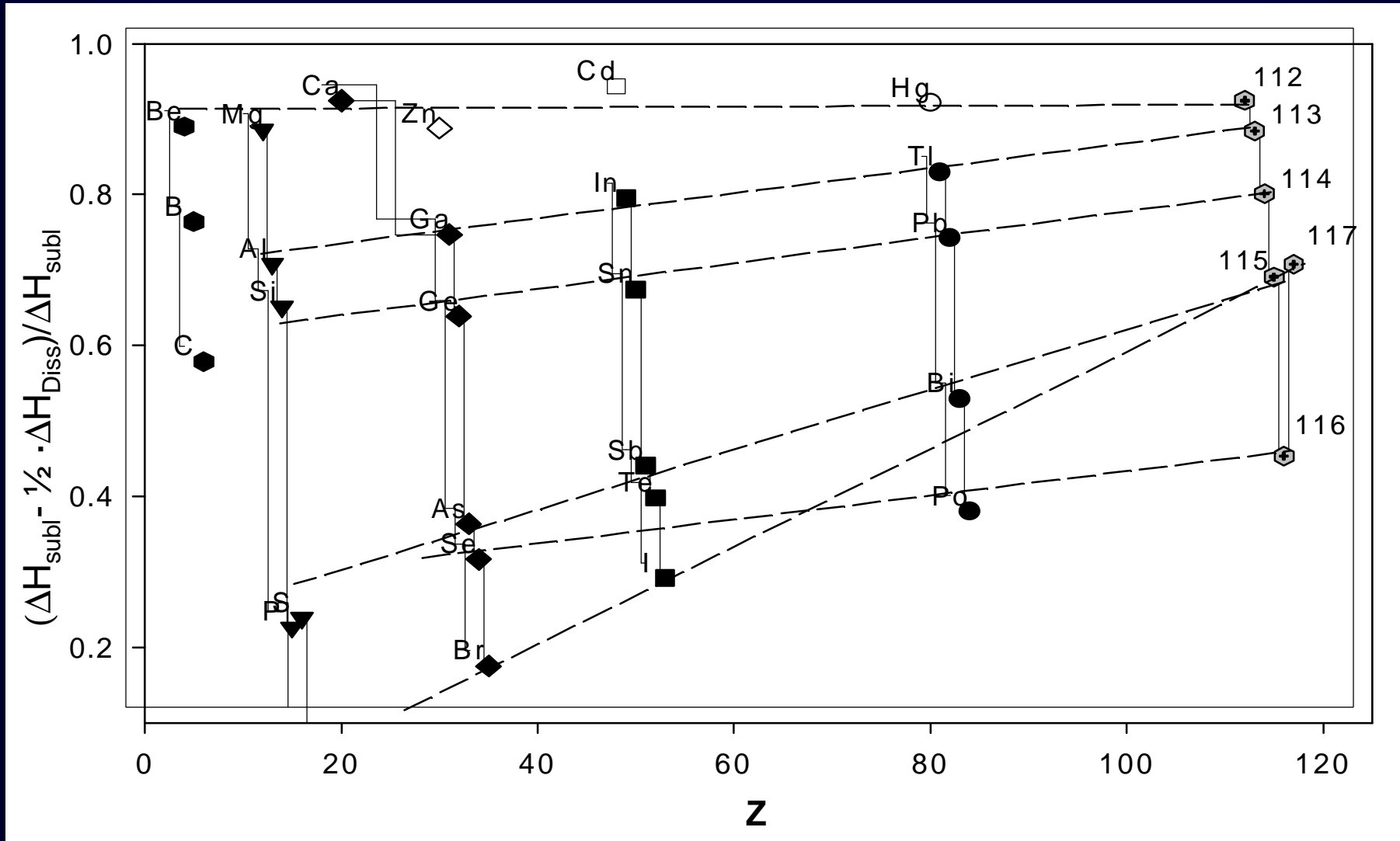
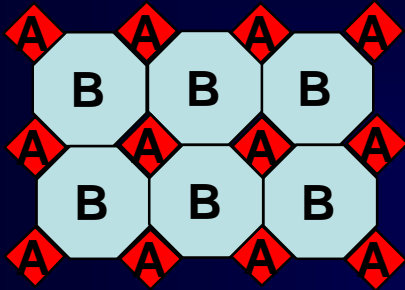


Transactinides

Metallic character



Miedema model: Intermetallic solution



$$\Delta H_{sol} = \frac{2 \cdot V_{Asol}}{n_{WSA}^{-1/3} + n_{WSB}^{-1/3}} \cdot \left(Q \cdot \left(n_{WSA}^{1/3} + n_{WSB}^{1/3} \right)^2 - P \left(\Phi_A^* - \Phi_B^* \right)^2 - R_m \right)$$

Semi empirical model adjusted to hundreds of binary compounds

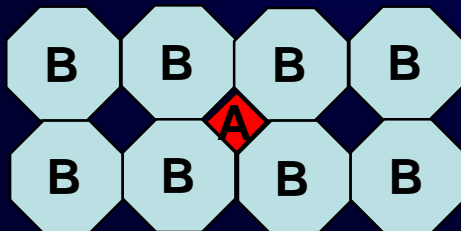
$$V_{Asol} = V_A \cdot \left(1 + a \cdot \left(\Phi_A^* - \Phi_B^* \right) \right)^{3/2}$$

A.R. Miedema, J. Less-Comm. Met. **46**, 67 (1975)

Eichler-Miedema model : Adsorption

In the surface:
Au

$$\Delta H_{ads} = -\Delta H_{subl} + 0.9 * \left(\Delta H_{sol} - \frac{V_{Asol}}{V_B} \cdot \Delta H_{vuf} \right) + \frac{V_{Asol}}{V_B} \cdot \Delta H_{svf}$$



$$\Delta H_{svf} = 0.0574 \cdot T_m(B)$$

$$\Delta H_{vuf} = 0.0803 \cdot T_m(B)$$

B. Eichler, Radiochim. Acta **33**, 121(1983)

Transactinides

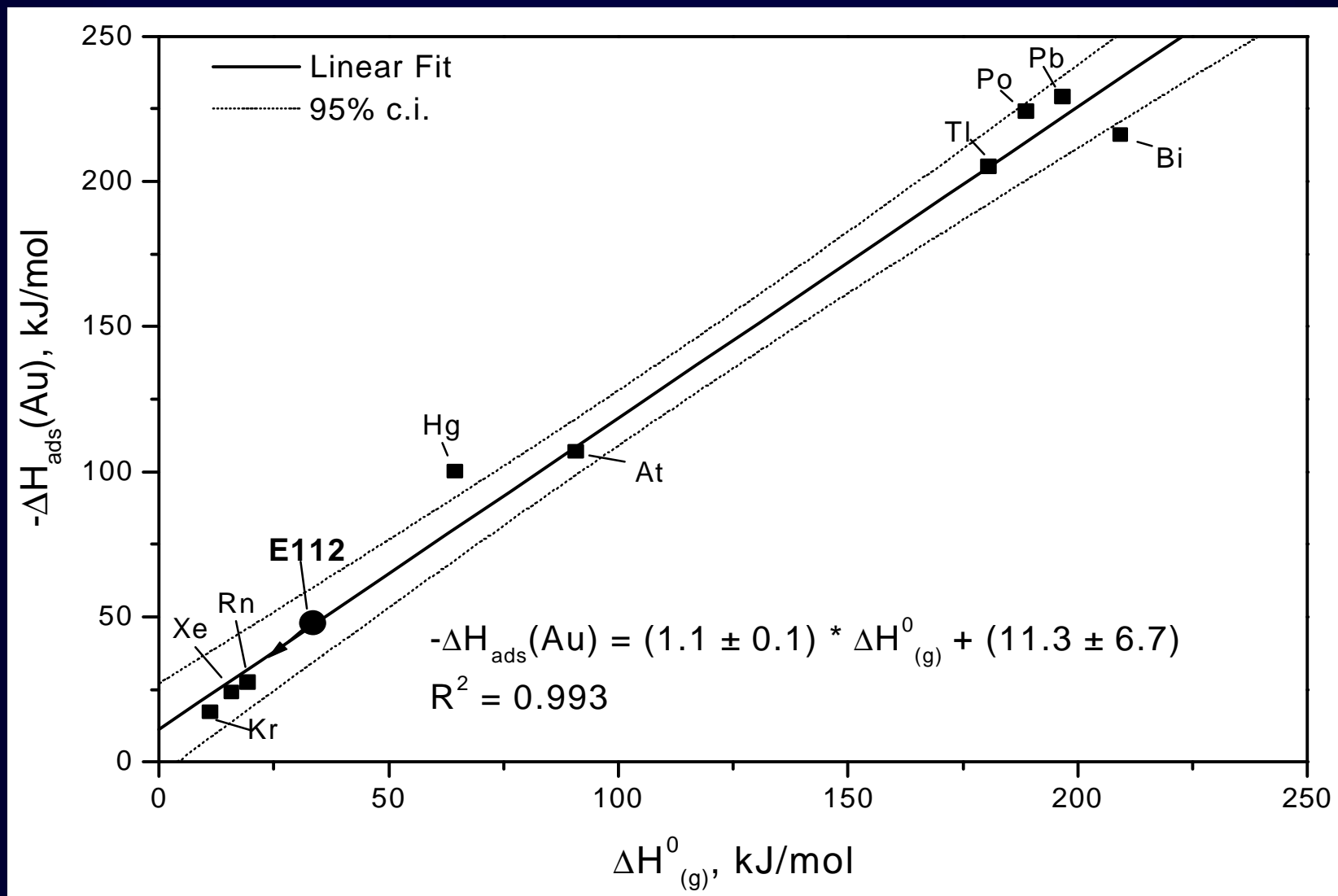
Eichler/Miedema Adsorption Enthalpies on Gold

Element	$\Delta H_{\text{subl}}^{[1]}$ kJ/mol	$\Delta H_{\text{ads}}^{\text{M}}(\text{Au})^{[2]}$ kJ/mol	$T_{\text{dep}}(\text{COLD})$ °C
112	39	-43	~ 25
113	234	-238	~ 600
114	190	-218	~ 500
115	184	-216	~ 500
116	146	-206	~ 500

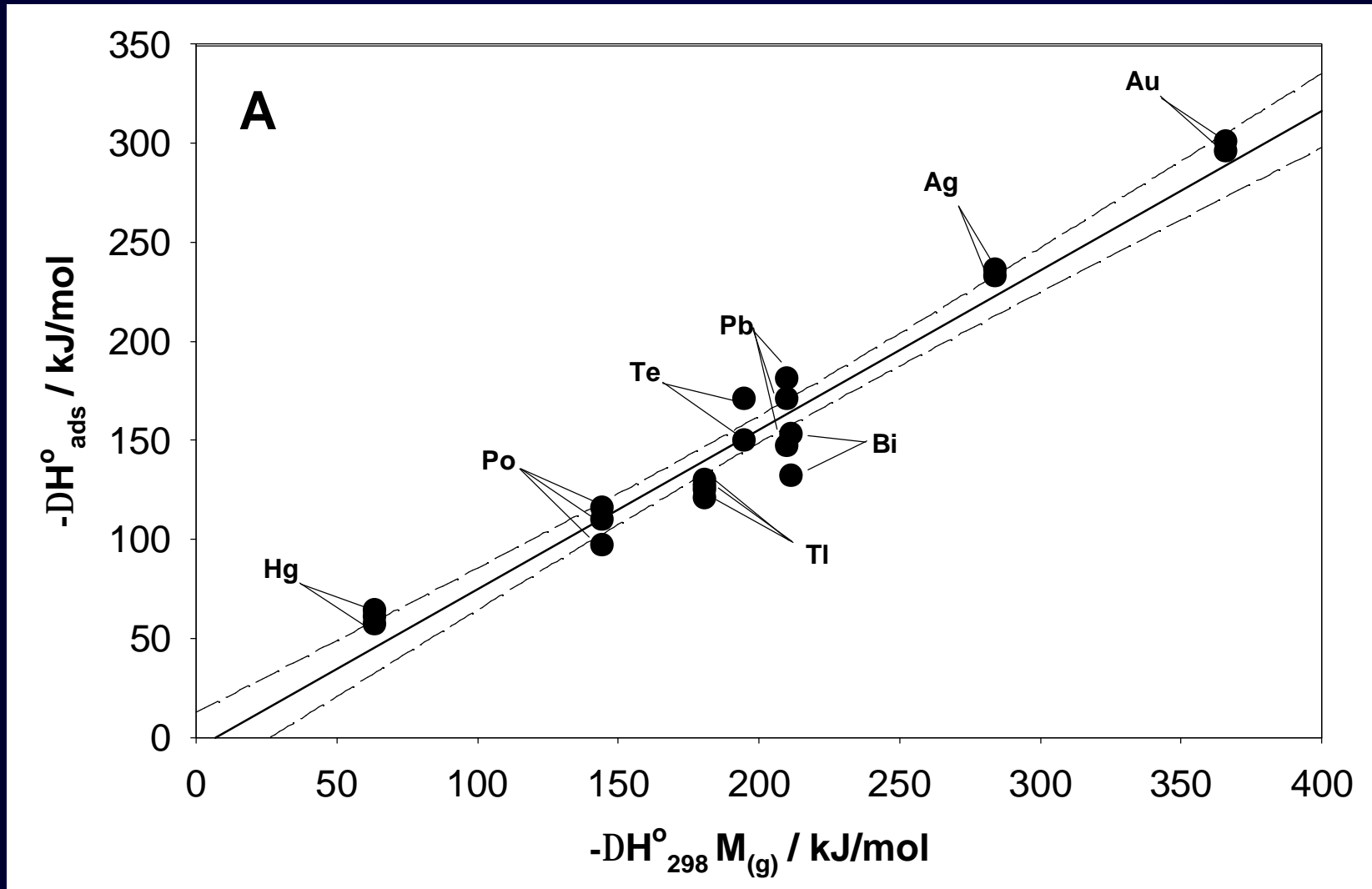
[1] B. Eichler, PSI Report 03-01, Villigen, 2003.

[2] B. Eichler, PSI Report 00-09, Villigen, 2000.

Adsorption on Gold



Adsorption on Quartz



Transactinides

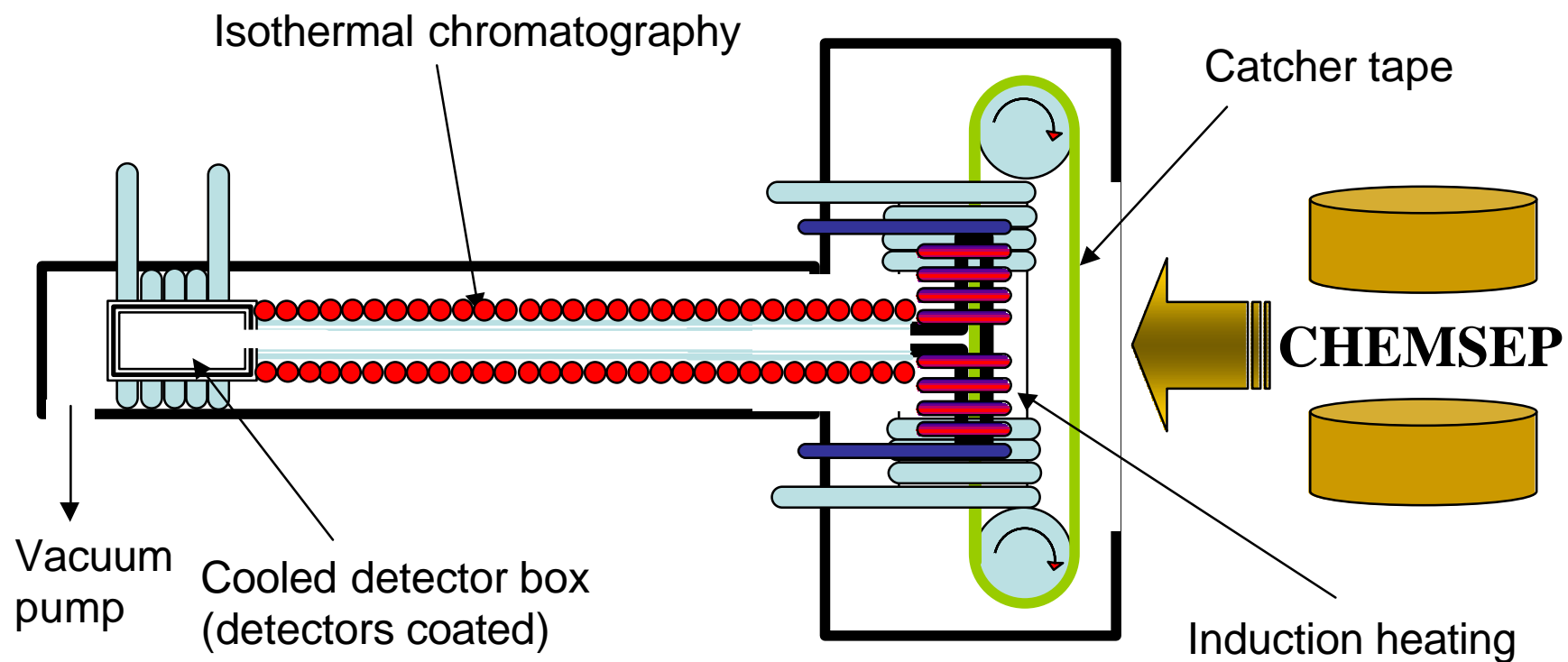
Adsorption enthalpies on quartz compared to gold calculated by use of correlations

Element	$\Delta H_{\text{subl}}^{[1]}$ kJ/mol	$\Delta H_{\text{ads}}(\text{Quartz})$ kJ/mol	$\Delta H_{\text{ads}}(\text{Au})$ kJ/mol
112	39	-25	-53
113	234	-182	-269
114	190	-148	-220
115	184	-143	-214
116	146	-112	-149
117	93	-68	-110
118	24	-13	-38

[1] B. Eichler, PSI Report 03-01, Villigen, 2003

Transactinides

Vacuum chromatography



Hot targets

Intermetallic actinide compounds with Rh

